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# **Structure Reports**

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# N-(4-Methoxyphenyl)quinoline-2-carboxamide

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#### **Key indicators**

Single-crystal X-ray study T = 294 KMean  $\sigma(C-C) = 0.002 \text{ Å}$ R factor = 0.042wR factor = 0.111 Data-to-parameter ratio = 16.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

The quinolyl and phenyl rings in the title compound,  $C_{17}H_{14}N_2O_2$ , are almost coplanar. There is an intramolecular hydrogen bond between the quinoline N atom and the amide N atom  $[N \cdots N = 2.6602 (17) \text{ Å and } N - H \cdots N = 112^{\circ}].$ 

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#### Comment

We have previously reported the structure of the bidentate ligand (R)-N-(1-phenylethyl)quinoline-2-carboxamide (Yang et al., 2001). The quinolyl and phenyl rings in that compound form a dihedral angle of 89.07 (5)°, while the corresponding rings in the title compound, (I), are coplanar. It is predicted that the N atom of the quinoline ring and the amide N atom or carbonyl O atom will coordinate to a metal ion and form a complex with a five-membered ring structure. There is an intramolecular hydrogen bond between the quinoline N atom and the amide N atom  $[N2 \cdots N1 = 2.6602 (17) \text{ Å}$  and N2—  $H2A \cdot \cdot \cdot N1 = 112^{\circ}$ ]. The structure of N-(4-iodophenyl)quinoline-2-carboxamide is reported in the following paper (Qi et al., 2003).

## **Experimental**

The title compound was synthesized from 2-quinolinecarboxylic acid and 4-methoxyaniline according to the general procedure of Johnson et al. (1960). The crystal used for the data collection was obtained by slow evaporation of a saturated DMF-H<sub>2</sub>O solution of (I) at room temperature.

Crystal data

 $C_{17}H_{14}N_2O_2$  $D_x = 1.340 \text{ Mg m}^{-3}$  $M_r = 278.30$ Mo  $K\alpha$  radiation Monoclinic,  $P2_1/n$ Cell parameters from 3598 a = 6.6372 (11) Åreflections b = 18.109 (3) Å  $\theta=1\text{--}27.5^\circ$  $\mu = 0.09 \text{ mm}^{-1}$ c = 11.4921 (19) Å $\beta = 92.867 (3)^{\circ}$ T = 294 (2) K $V = 1379.6 (4) \text{ Å}^3$ Prism, colorless  $0.38 \times 0.28 \times 0.26 \text{ mm}$ 

Data collection

Siemens SMART CCD area-3167 independent reflections detector diffractometer 1829 reflections with  $I > 2\sigma(I)$  $\varphi$  and  $\omega$  scans  $R_{\rm int} = 0.034$ Absorption correction: multi-scan  $\theta_{\rm max}=27.5^\circ$ (SADABS; Sheldrick, 1996)  $h = -8 \rightarrow 8$  $T_{\min} = 0.967, T_{\max} = 0.977$  $k = -13 \rightarrow 23$ 9204 measured reflections  $l = -14 \rightarrow 14$ 

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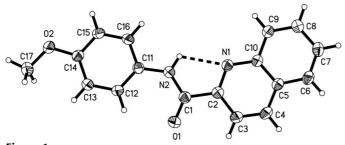


Figure 1
The molecular structure of (I), showing ellipsoids at the 30% probability level (Siemens, 1995).

#### Refinement

Refinement on  $F^2$   $w = 1/[\sigma^2(F_o^2) + (0.05P)^2]$   $R[F^2 > 2\sigma(F^2)] = 0.042$   $where <math>P = (F_o^2 + 2F_c^2)/3$   $wR(F^2) = 0.111$   $(\Delta/\sigma)_{\max} < 0.001$  S = 1.01  $\Delta\rho_{\max} = 0.15 \text{ e Å}^{-3}$  3167 reflections  $\Delta\rho_{\min} = -0.17 \text{ e Å}^{-3}$  Extinction correction: SHELXL97 H-atom parameters constrained Extinction coefficient: 0.020 (2)

**Table 1** Hydrogen-bonding geometry (Å, °).

$D-H\cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-\mathrm{H}\cdots A$
N2−H2 <i>A</i> ···N1	0.86	2.21	2.6602 (17)	112

The C-bound H atoms were placed in geometrically calculated positions and included in the final refinement in the riding-model approximation. The H atom on N2 was initially refined but was constrained in the final refinement.

Data collection: *SMART* (Siemens, 1995); cell refinement: *SMART*; data reduction: *SHELXTL-NT* (Siemens, 1995); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *SHELXTL-NT*; software used to prepare material for publication: *SHELXTL-NT*.

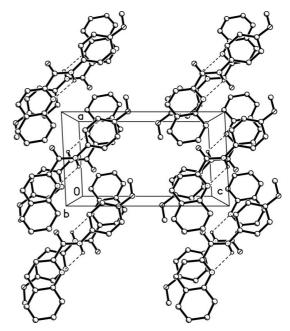


Figure 2 The molecular packing along the b axis.

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## References

Johnson, W. A., King, T. J. & Turner, J. R. (1960). J. Chem. Soc. pp. 1509–1511.
Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
Sheldrick, G. M. (1997). SHELXS97 and SHELXTL97. University of Göttingen, Germany.

Qi, J. Y., Qiu, L. Q., Yang, Q. Y., Zhou, Z. Y. & Chan, A. S. C. (2003). Acta Cryst. E59, o104–o105.

Siemens (1995). SMART (Version 5.0) and SHELXTL-NT (Version 5.10). Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA. Yang, Q. Y., Zhou, Z. Y. & Qi, J. Y. (2001). Acta Cryst. E57, 0971–0972.